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concentration on the  
rate of a chemical  
reaction. (OR The  
objective of this lab is  
to observe the effect of  
temperature and  
concentration on the  
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reaction.) Procedure:

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Please refer to Heath  
Chemistry: Laboratory  
Experiments (Canadian  
Edition), DiSpezio,  
Michael, A., et al, 1987,  
D.C. Heath Canada,  
Ltd. pg. 197-203

**A Sample Lab Report  
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**Introduction**

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This lab is an excerpt from Heath Chemistry, a textbook of laboratory experiments.

Information regarding the lab and detailed procedure are provided on the following pages.

Note: Change Step 2 from g. 0 g of copper (II) chloride to approximately 3.0 g - 4.0 g

**Lab 4B (student**  
*Page 18/23*

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Chemistry lab safety rules. As chemistry labs are one of the most common types, these basic chemistry lab safety rules are relevant to many scientists, dealing with the safe performance of common activities and tasks in the average chemistry lab: Before you start an experiment, make sure you are fully aware of the hazards of the

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**Lab Safety Rules  
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19A. Chemistry 12 -  
Lab 19A Flow Chart  
(Investigating Chemical  
Equilibrium) PART 1:  
Put on safety goggles  
and lab apron Obtain 2  
(Empty, Clean, AND  
DRY) 250mL  
Erlenmeyer flasks. Add  
approx. 100mL of  
DISTILLED water and  
1mL of thymol blue  
solution to each flask

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Record the colour of this solution in note book FIRST FLASK: add a single drop of 0.1M HCl.

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